

# Some Physicochemical Peculiarities of Poplar Plastocyanins *a* and *b*

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The redox potentials of poplar plastocyanins *a* and *b* (PC*a*, PC*b*) were determined by spectrophotometric titrations of their reduced forms with  $[\text{Fe}(\text{CN})_6]^{3-}$ . It was found that the two isoforms have the following millimolar extinction coefficients  $\epsilon_{597}$ , equilibrium constants  $K_{\text{eq}}$  of one-electron exchange with  $[\text{Fe}(\text{CN})_6]^{4-}/[\text{Fe}(\text{CN})_6]^{3-}$ , and standard electron potentials  $E^{0\text{R}}$ :

PC*a*:  $\epsilon_{597} = (4.72 \pm 0.08) \text{ mM}^{-1} \text{ cm}^{-1}$ ,  $K_{\text{eq}} = 0.133 \pm 0.009$ ,  $E^{0\text{R}} = (354 \pm 11) \text{ mV}$ ;

PC*b*:  $\epsilon_{597} = (5.23 \pm 0.16) \text{ mM}^{-1} \text{ cm}^{-1}$ ,  $K_{\text{eq}} = 0.175 \pm 0.010$ ,  $E^{0\text{R}} = (363 \pm 12) \text{ mV}$ .

The pH dependence of the redox potential of PC*b* was studied too. It was found, that the value of  $E^{0\text{R}}$  for PC*b* is constant in the pH range 6.5–9.5, but decreases in the range 4.8–6.5. On the whole, the dependence resembles that of PC from some well-known plant species, including poplar PC*a*. The changes of  $E^{0\text{R}}$  in the pH-dependent region for poplar PC*b*, however, are smaller and are 13 mV per pH unit, whereas in the other well-known plant species the changes are about 50–60 mV per pH unit. It has been assumed that the weaker pH dependence of  $E^{0\text{R}}$  of PC*b* accounts for some structural differences between PC*a* and PC*b*.

**Key words:** Plastocyanin, Dimorphism, Photosynthesis